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(17) Applicant: ETHICON INC.  
U.S. Route 22  
Somerville New Jersey 08876(US)

(18) Inventor: Hinech, Bernhard, Dr.  
Travestrasse 1  
D-2000 Norderstedt(DE)  
Inventor: Walther, Christoph, Dr.  
AM Brahmberg 19  
D-2358 Kattendorf(DE)

(19) Representative: Jones, Alan John et al  
CARPMAELS & RANSFORD 43 Bloomsbury  
Square  
London, WC1A 2RA(GB)

(54) Implant

(57) The invention relates to an implant from an open-cell, foam-like plastic material based on resorbable polyesters, such as poly-p-dioxanone, other polyhydroxycarboxylic acids, polylactides or polyglycolides, as well as their copolymers, in which one or more textile reinforcing elements made from resorbable plastic are embedded in an open-cell plastic matrix with a pore size of 10 to 200  $\mu\text{m}$ .

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being given in the subclaims. In addition, for solving the inventive problem, processes for producing the implants are proposed.

The invention is based on the surprising finding that the tensile strength of such implants is dependent on the polymer content of the solution to be freeze dried and is only dependent to a reduced extent on the production conditions adopted during freeze drying, such as the freezing rate. However, a part is also played by the choice of solvent and this also influences the pore size distribution, the pore size and the pore shape.

It has surprisingly been found that on incorporating textile reinforcing elements of resorbable plastic such as fibres, yarns, braids, knitted fabrics and the like, it is possible to significantly improve the mechanical strength, without modifying the porosity characteristics, flexibility and elasticity of the foam materials. As a function of the choice of reinforcing elements, the mechanical strength can be increased in one or more directions in space, in that e.g. parallel fibres or threads increase the stability in only one direction and net-like flat structures improve the same in all directions of the corresponding surfaces.

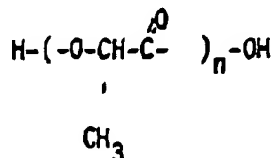
The invention is further illustrated hereinafter by means of examples

#### 1. Production of the Plastic Matrix

The production of the starting polymers is described hereinafter:

##### Example A: Polylactides

The starting polymer is produced in that purified lactide crystals are polymerized with 0.002% by weight of tetraphenyl tin, whilst excluding oxygen, for 4 hours in a temperature range of 150 to 200°C, the polymer obtained is precipitated from dioxan by adding water and then dried under a high vacuum. These polylactides have the following general formula:

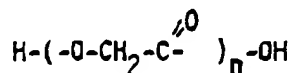


in which n has a value of 300 to 1200 corresponding to a molecular weight of 20 000 to 85 000.

Poly-L-lactide, poly-DL-lactide, as well as copolymers of these two substances and copolymers with glycolides can in particular be used for the inventive products and processes. Apart from glycolides, the polylactides can also contain other monomers, such as are referred to e.g. in US patent 4 181 983, column 3, lines 26 to 36

##### Example B: Polyglycolides

The inventively used polyglycolides are obtained by the polymerization of glycolic acid and have the following general formula:



in which n is such that the molecular weight is in a range 10 000 to 500 000. These polyglycolides are prepared in that glycolide is polymerized with tin (II)-octanoate and glycolic acid, whilst excluding oxygen, and in vacuo, for approximately 4 to 5 hours at 185 to 230°C and after cooling the polyglycolide is isolated as a white, opaque, viscous material. The inherent viscosity, measured in 1,1,1,3,3,3-hexafluoro-2-propanol is in the range 0.5 to 4.

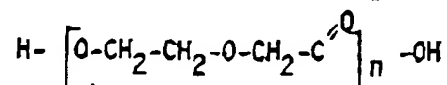
Example C: Copolymers of Glycolide and Lactide

8 parts by weight of glycolide and 1 part by weight of L-lactide are polymerized with glycolic acid and tin (II)-caprylate, as described in DE-OS 21 62 900, whilst excluding oxygen and moisture for approximately 4 hours at 200°C. The inherent viscosity of the polymer is 0.5 to 2.5 when measured in 1,1,1,3,3,3-hexafluoro-2-propanol.

Block copolymers of glycolide and lactide can also be used for the inventive products and processes, as described in DE-OS 28 49 785.

Example D: Polydioxanone

The inventively used polydioxanone has the following general formula



Polydioxanone is polymerized by the ring-opening polymerization of 2-oxo-1,4-dioxan in the presence of Sn catalysts, such as e.g. tin (II)-caprylate at 120 to 150°C, whilst excluding oxygen and moisture, within 1 to 4 hours. The inherent viscosity (measured in 1,1,1,3,3,3-hexafluoro-2-propanol) of the polymer is in the range 0.5 to 3 and preferably 1.5 to 2.2.

II. Preparation of the Plastic Solution or Dispersion

Decisive for the preparation of the plastic matrix are the solvent chosen and the concentration of the resorbable plastic in the solvent to be freeze dried and optionally, in the case of a modified process, the addition of crystalline additives. Other solvents can be used, particularly in the latter case.

A. Solvent influence

According to the invention use is made of the following solvents:

- a) 1,4-dioxan
- b) 1,4-dioxan with an acetate of a C<sub>2</sub> to C<sub>3</sub> alcohol, namely ethyl acetate in the ratio 9:1.
- c) hexafluoroisopropanol and as the comparison solvent
- d) benzene with and without surfactant.

In each case 5 g of a poly-L-lactide according to example A are introduced into 100 g of the particular solvent and frozen from 20°C to -60°C in between 30 and 300 seconds and then freeze dried in vacuo.

The following table I shows the influence of the solvent selection on the pore structure of the polymer matrix.

TABLE I

Influence of the Solvent Selection on the Pore Structure

Solvent	Pore Characterization
A-a dioxan	open, round pores ; diameter 20-50 $\mu\text{m}$
A-b dioxan/ethylacetate 9:1	open, round pores ; diameter 20-60 $\mu\text{m}$
A-c hexafluoroisopropanol	open, round pores, fibrous ; diameter 10-50 $\mu\text{m}$
A-d <sub>1</sub> benzene	long channels with a diameter under 10 $\mu\text{m}$ , a few oval pores of 20 - 100 $\mu\text{m}$
A-d <sub>2</sub> benzene + 2.5% surfactant	elongated pores (fibrous product); diameter 5-25 $\mu\text{m}$

The values of table I clearly show that only the inventively used solvents lead to a suitable polymer matrix. Similar results were obtained with the other polymerizable polymers according to example B (polyglycolides), example C (copolymers) and example D (polydioxan), as is shown by table II.

B. Influence of the Concentration

Different polymers in different concentrations were treated in a solvent mixture of dioxan and ethyl acetate in a ratio of 9:1 as in A-b) and freeze dried. The following table III shows that excellent pore sizes and distributions are obtained on respecting a concentration in the middle of the inventive concentration range.

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TABLE II

Influence of the Solvent Selection on the Pore Structure of

Copolymers of L-Lactide and Glycolide

Glycolide (Weight Ratio in the Copolymer)	Lactide	Solvent	Pore Characterization
90	10	HFIP	Fibrous, round, open pores. diameter approximately 20 - 100 $\mu\text{m}$
30	70	dioxan/ethylacetate 9:1	Round, open pores with a diameter of 20 - 50 $\mu\text{m}$

TABLE III

## Influence of the Polymer Concentration on the Pore Size

Polymer	Concentration (g/100 ml solvent)	Pore Characterization
Poly-L-lactide	5	open, round pores; diameter 20- 50 $\mu$ m
	7.5	open, round pores; diameter 20-100 $\mu$ m
	12	open, round pores; 80% of the pores 20-100 $\mu$ m 20% of the pores $\leq$ 20 $\mu$ m
	16	open, round pores; 10% of the pores 20- 30 $\mu$ m 90% of the pores $\leq$ 20 $\mu$ m
Copolymer of glycolide and lactide 3:7	12	open, round to elongated pores; 80% of the pores 20-100 $\mu$ m 20% of the pores < 20 or > 100 $\mu$ m
	7.5	open, round to elongated pores; diameter 20- 80 $\mu$ m
Copolymer of glycolide and lactide 3:7	7.5	open, round pores; 88% of the pores 20-100 $\mu$ m 10% of the pores $\leq$ 20 $\mu$ m 2% of the pores 100-150 $\mu$ m
	7.5	



### C. Adjusting the Pore Size Distribution by Crystalline Additives

4 parts of a copolymer of L-lactide and glycolide (8:1) and 1 part of citric acid are pulverized, screened and classified, the mixture being dissolved and/or suspended with 40 parts of 1,1,1,3,3,3-hexafluoro-2-propanol. As described in A, this mixture is frozen and freeze dried.

The citric acid is extracted with tetrahydrofuran from the resulting crude foam.

The foam contains circular, open pores, whose pores sizes are largely dependent on the particle size distribution of the citric acid used. An example is given in table IV.

TABLE IV

Influence of Crystalline Additives  
on the Pore Structure

Particle Size	Citric Acid	Pore Size in Foam
80%	20 $\mu$ m to 40 $\mu$ m	75% 70 $\mu$ m to 40 $\mu$ m
80%	$\leq$ 20 $\mu$ m	90% $\leq$ 20 $\mu$ m

### III. Production of the Implant

#### Example 1

Different solutions of 2.0 to 30.0 g of poly-L-lactide are in each case dissolved in 100 ml of 1,4-dioxanethylacetate (90:10). A reinforcing element of the copolymer of L-lactide and glycolide in the weight ratio of 8:1 is in each case introduced into these solutions. The net in a dish is then frozen for 30 to 90 seconds to -60°C and then freeze dried.

Table V shows the value of the tensile strength of the polymer matrix without reinforcing inserts, determined with a tensiometer at 50 mm/min and a 2x10x50 mm test piece, the tensile strength through the surface of fracture gave the indicated tensile strength.

The tests revealed a rise in the tensile strength of the polymer matrix with increasing polymer concentration (no. 1-4) and under lower freezing conditions (test 5), as well as the comparatively inferior values in the case of a conventional solvent (tests 2 and 6 compared with test 7).

TABLE V

Tensile Strength of Porous Implants Without Reinforcing Elements

Test No.	Polymer Concentration (g/100ml of solvent)	Tensile Strength N/mm <sup>2</sup>	Freezing Conditions	Solvent
1	4	0.10	A	1,4-dioxan/ethyl acetate 7:1
2	6	0.35	A	
3	10	0.90	A	
4	15	1.95	A	
5	4	0.28	B	
6	6	0.53	A	1,4-dioxan/ isoamyl acetate 19:1
7	6	0.30	A	benzene

A 20°C to -60°C in 30-300 seconds

B 20°C to -130°C in 30-300 seconds

Table VI shows the surprising increase in the tensile strength when using textile reinforcing elements in a polymer matrix according to example 1.

TABLE VI

Tensile Strength of Reinforced Porous Implants

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	Polymer Concentration (g/100ml of solvent)	Tensile Strength N/mm <sup>2</sup>	Reinforcing Element
10	4	0.10	without
15	4	1.35	2dpf yarn, length 2-5 cm
	4	8.0	knitted net
20	10	1.30	2dpf yarn, 1-5 mm
	10	10.2	knitted net

Table VII shows the increase in the tensile strength of porous tubular implants with woven reinforcing elements of different weaving structures.

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TABLE VII

Breaking Force of Porous Implants with Woven Reinforcing Elements

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Breaking Force	Reinforcing Element
1.4N	none
40	thick-walled hose, 2x56 denier threads
70	impervious hose, 4x56 denier threads
150	very thick-walled, impervious hose, 8x56 denier threads

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Test pieces: external diameter 4 mm  
internal diameter 2.7 mm

Polymer concentration: 5g/100ml of solvent

Reinforcing element: copolymer of L-lactide and glycolide 1:9

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**Claims**

1. Implant of an open-cell, foam-like plastic material based on resorbable polyesters such as poly-p-dioxanone, other polyhydroxy carboxylic acids, polylactides or polyglycolides, as well as their copolymers, characterized in that one or more reinforcing elements of a textile nature formed from resorbable plastic are embedded in an open-cell plastic matrix with a pore size of 10 to 200  $\mu$ m.

2. Implant according to claim 1, characterized in that compared with the resorbable plastic of the matrix, the resorbable plastic of the textile reinforcing elements have the same or a slower resorbability.

3. Implant according to claim 1, characterized in that the textile reinforcing elements are knitted, woven, twisted, braided or as felts in the form of fibres, threads, hoses, strips or fleeces.

4. Implant according to claims 1 to 3, characterized in that the open-cell plastic matrix has a pore size of 20 to 150  $\mu\text{m}$ .

5. Implant according to claims 1 to 3, characterized in that the density of the plastic matrix is 0.05 to 0.80  $\text{g/cm}^3$ .

6. Process for the production of an implant according to claims 1 to 5, characterized in that poly-p-dioxanone, polylactides or polyglycolides are dissolved in a solvent in a concentration of 5 to 30 parts by weight of polymer and the textile reinforcing element is frozen in a mould together with the plastic solution and then the solvent is removed by freeze drying.

7. Process according to claim 6, characterized in that the solvent used is hexafluoroisopropanol, 1,4-dioxan or a mixture of 1,4-dioxan and an acetate of a  $\text{C}_2$  to  $\text{C}_4$  alcohol in a volume ratio of 99:1 to 50:50.

8. Process according to claims 6 and 7, characterized in that use is made of poly-p-dioxanone with an inherent viscosity of 0.5 to 3.0 or polylactides with an inherent viscosity of 0.5 to 2.2 or polyglycolides with an inherent viscosity of 0.5 to 4.0.

9. Modification of the process according to claims 6 to 8, characterized in that the textile reinforcing element is impregnated with the plastic solution and then frozen and freeze dried.

10. Modification of the process according to claims 6 to 8, characterized in that crystalline organic compounds, salts of organic acids or inorganic salts are added to the solvents and after freeze drying are extracted from the implant with a suitable inert solvent.